

Cambridge Assessment International Education

Cambridge International General Certificate of Secondary Education

PHYSICAL SCIENCE 0652/32

Paper 3 Core Theory

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MARK SCHEME
Maximum Mark: 80

Published

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Question	Answer	Marks
1(a)(i)	measuring cylinder;	1
1(a)(ii)	balance;	1
1(b)(i)	154 (g);	1
1(b)(ii)	use of $\rho = m/v = (154/118)$;	3
	= 1.3 ;	
	g/cm³;	
1(c)(i)	expands;	2
	decreases;	
1(c)(ii)	convection;	1

Question	Answer	Marks
2	hydrocarbons ;	5
	fractional distillation ;	
	boiling point;	
	bitumen ;	
	(addition) polymerisation ;	

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Question	Answer	Marks
3(a)	visible / ultraviolet ;	3
	Sun;	
	electrical (potential);	
3(b)(i)	splitting of nucleus ;	3
	into two new nuclei ;	
	detail e.g. initial nucleus large / mention of other fission fragments ;	
3(b)(ii)	less carbon dioxide / carbon monoxide / sulfur dioxide / eq ;	2
	danger of radioactive material leaking / cost of construction or decommissioning / reprocessing (of spent fuel rods) / eq;	

Question	Answer	Marks
4(a)(i)	oxygen is 21% of air;	1
4(a)(ii)	nitrogen / N ₂ ;	1
4(b)(i)	4 3 2 ;	1
4(b)(ii)	oxidation;	1
4c	no reaction / no change (in water level);	2
	copper is unreactive / less reactive than iron / does not react with oxygen / eq;	
5(a)(i)	i = r or equal;	1
5(a)(ii)	ray bends at BOTH boundaries ;	2

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Question	Answer	Marks
	ray refracted towards, but not beyond normal AND emergent ray parallel to incident ray ;	
5(b)(i)	critical (angle);	1
5(b)(ii)	reflected back into water $i = r$ (approx.);	1

Question	Answer	Marks
6(a)(i)	use a tap funnel / longer thistle funnel reaching into the acid;	1
6(a)(ii)	any two from:	2
	heat more concentrated acid increase surface area of zinc ;;	
6(a)(iii)	pops / explodes / burns with a blue flame ;	1
6(b)(i)	substance used to produce (heat / thermal) energy;	1
6(b)(ii)	water / H ₂ O ;	1
6(b)(iii)	any two from:	2
	difficult to transport (as a gas) / (liquid) petrol is easier to transport	
	difficult to store / needs pressure vessel to store / needs to be at high pressure to store	
	not readily available	
	other sensible suggestions ;;	

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Question	Answer	Marks
7(a)	rock containing metal;	1
7(b)(i)	background radiation / radiation from surroundings ;	1
7(b)(ii)	B; E;	2
7(b)(iii)	any two from:	2
	helium nucleus	
	two protons and two neutrons	
	fast moving / double electronic charge / positively charged particle ;;	
7(b)(iv)	random (nature of emissions) ;	1

Question	Answer	Marks
8	gas; VII / 7 / seven; gas; 2.8.1; Na+; 16; 32;	7

Question	Answer	Marks
9(a)(i)	variable resistor ;	1
9(a)(ii)	(current) decreases;	1
9(b)(i)	correct circuit symbol ;	2
	connected in parallel with the battery ;	

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Question	Answer	Marks
9(b)(ii)	use of $V = I R$ or $I = V/R = 4.6 \div 8.0$;	2
	0.58 or 0.575 ;	
9(c)(i)	Z correctly drawn in parallel with Y;	1
9(c)(ii)	greater;	1

Question	Answer	Marks
10(a)(i)	found uncombined / as metallic element / OWTTE ;	1
10(a)(ii)	silver / mercury / any correct ;	1
10(b)(i)	bauxite ;	1
10(b)(ii)	(aluminium is) more reactive than carbon / higher in reactivity series ;	1
10(b)(iii)	2 correct uses of stainless steel cutlery / surgical instruments / exhaust pipes / eq. ;;	2

Question	Answer	Marks
11(a)	B A all three correct;;	2
11(b)(i)	poles correct;	2
11(b)(ii)	(like poles) repel;	1

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Question	Answer	Marks
12(a)	any two from:	2
	(a family of) similar compounds	
	with similar properties	
	same functional group	
	same general formula ;;	
12(b)	H	1
	H-C-	
	H ;	
	rest of molecule correct;	1
12(c)	water / H ₂ O ;	1
12(d)(i)	any number in range 3 to below 7;	1
12(d)(ii)	Universal indicator / pH meter ;	1

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